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Organometallic species that are bridged by organic groups can also serve as Lewis acids and, in the process, bridge cleavage can take place.

$$\text{Al}_2(\text{CH}_3)_6 + 2\text{N}(\text{C}_2\text{H}_5)_3 \rightarrow 2(\text{CH}_3)_3\text{AlN}(\text{C}_2\text{H}_5)_3$$

Electron deficient organometallic species are Lewis acids.

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