

Modern Chemistry Chapter 5 Review Answer Key

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Modern Chemistry Chapter 5-Final Review. STUDY. PLAY. Periodic law. The physical and chemical properties of the elements are periodic functions of their atomic numbers. periodic table. An arrangement of the elements in order of their atomic numbers so that elements with similar properties fall in the same column, or group.

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Modern Chemistry 37 The Periodic Law CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. ____ When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released. 2.

CHAPTER 5 REVIEW The Periodic Law

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered.

5 The Periodic Law

Chemistry: Chapter 5 Review. Describe.... 1) Dalton's model - solid sphere, made up of only 1 type of matter, no space. 2) Schrodinger's model - based on probability of finding an e-; electrons are found in orbitals (look like electron clouds), which are in sublevels (s,p,d,f,g), which are in energy levels.

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Chapter 5 Chemistry review. electromagnetic radiation. Wavelength. frequency. Amplitude. a kind of radiation including visible light, radio waves, gamm.... Horizontal distance between the crests or between the troughs.... the number of complete wavelengths that pass a point in a give.... Height of a wave.

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