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1. Introduction to Thermodynamics

Shown to the right is a PV diagram with two adi-abetic lines 1 → 2 and 2 → 3, assumed to intersect at point 2. A cycle is formed by an isothermal line from 3 → 1. An engine traversing this cycle would produce work. For the cycle ΔU = 0, and therefore by the first law, Q + W = 0. Since W is negative, Q must be positive, indicating that heat is absorbed by the system.

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