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a measure of how much of a given substance is mixed with another substance. Diffusion. the spontaneous net movement of particles from an area of high concentration to an area of low concentration. Dynamic equilibrium. a state of balance in

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which there is little or no total change.

Osmosis. the movement of solvent molecules across a semipermeable membrane from an area of high solvent concentration to an area of low solvent concentration.

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and observe the
system as it adjusts to
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concentration of the
solute can be
manipulated, along

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with the volume of the cell.

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Osmosis is one of the methods through which solvents. This

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movement involves the transfer of water from a less concentrated solution to a more concentrated solution through a semipermeable membrane. Most people confuse it with diffusion, which does not involve movement through a membrane. See how much you understood about osmosis in the class by taking the test below before the test.

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Chemistry Quiz On Osmosis! Trivia Questions - ProProfs Quiz

It is a semipermeable membrane, which means that some particles pass through the membrane easily while others cannot.

The Osmosis Gizmo™ portrays a cell (red square) in a solution of purple solute particles dissolved in green solvent particles. Press

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Play () and observe. 1.

OsmosisSE - Name **Date Student** **Exploration Osmosis**

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