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Gas Table 20.1 Densities of Various Gases Density (kg/m3)\* Dry air 0° C 10° C 20° C 30° C Helium Hydrogen Oxygen \* At sea level atmospheric pressure and at 0° C (unless otherwise specified) 1.29 1.25 1.21 1.16 0.178 0.090 1.43 Table 20.1 shows how the density of air changes with tempera-ture. At sea level, 1 cubic meter of air at 20°C has a mass of about 1.2 kg.

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### Exercises

152 Conceptual Physics Reading and Study Workbook N Chapter 19 19.2 Buoyancy (pages 366–367) 10. The is the net upward force exerted by a fluid on a submerged or immersed object. Match each sentence with the correct result. 11. The weight of a submerged object is greater than the buoyant force. 12. The weight of a submerged object is less ...

### Exercises - PHYSICS Mr. Bartholomew

of a gas. "Pressure svolume" for a sample of gas at one time is equal to any "different pressure sdifferent volume" of the same sample of gas at any other time. In equation form, P 1V 1 = P 2V 2. 20.6 Buoyancy of Air Any object less dense than the air around it will rise. • Archimedes' principle for air states that an object ...

### Summary - University Homepage

Conceptual Physics, 11e (Hewitt) Chapter 14 Gases and Plasmas 14 Conceptual physics chapter 14 gases and plasmas review question answers. 1 Questions About Gases 1) The energy source responsible for the motion of molecules that make up the atmosphere is A) their own natural kinetic energy. B) the pressure caused by the weight of air. C) atmospheric tides.