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Page 29 and 30: 29 Information: Average Atomic Mass. Page 31 and 32: 31 ChemQuest 10 Name: _____ Page 33 and 34: 33 An electron that absorbs energy . Page 35 and 36: 35 Information: Orbitals So far we ; Page 37 and 38: 37 ChemQuest 12 Name: _____ Page 39 and 40: 39 Critical Thinking Questions 6. H

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ChemQuest 26. Name: _____ Date: _____ Hour: _____ Information: Definition of Electronegativity. Electronegativity. is a measure of how much an atom attracts an electron. The higher the electronegativity, the greater the atom's attraction for electrons. Atoms that become negative ions have a much greater electronegativity than atoms that ...

ChemQuest 26

ChemQuest 37 SslgDcll?g E"a1ue Date Hour l ntroduction Questions Information: Subscripts and Coefficients A subscript is a smail number that tells you how many atoms are in a compound.For exampie, in CaCl₂ the two is the subscript and it tells us that there are two chloride ions bonded to one calcium. A coefficient tells also tells us how many atoms or compounds there are, but in a different way.

SslgDcll?g Ea1ue - LTHS Answers

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Classwork and Homework Handouts

Transcript Chemquest 7 - Wayne County Public Schools ... $\square\square$ Mass does not change, weight is affected by the amount of gravity pulling down on an object. SKILLS PRACTICE 7 1. 29.6 g/cm³ 2. 19.2 mL 3. 7.6 x 10¹¹ cm³ 4. 0.0028 kg/cm³ 5. 1,200,000 g/L 6. 0.884 g/cm³ 7. 38.15 mL Company; Nicosia Constantinou Palaiologou 16, Palouriotissa, 1040 ...

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