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100 dm³ of the contaminated drinking water was reduced by boiling, to 7.50 dm³. It was found that when the reduced volume was tested it had an absorbance of 0.55. Calculate the concentration of lead ions (in mg dm⁻³) in the original contaminated

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3. a ionic 1 3. b i red 1 3. b ii emission spectra of both ${}^6\text{Li}$ and natural Li give same colour/produce same «range of» wavelengths OR they have same electron transitions/same nuclear charge Accept “the spectra are almost identical”. 1 3. b iii ICP -MS Accept “MS/mass spectrometry”. 1 3. c r m 0.694 n 0.100 mol M 6.94 «

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point compared to oleic acid, $\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$. Outline your reason. [2]..... B3. (a) Some synthetic hormones are similar in structure to progesterone and estrogen and may be used to prevent pregnancy. Outline the mode of action of these hormones as oral contraceptives. [3].....

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3.0 2.0 1.0 Chemical shift / ppm (a)
Identify what information from the

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spectrum allows the determination of the relative numbers of hydrogen atoms producing each peak. [1]..... (b) Deduce which of the following compounds is Q.
[1] $\text{CH}_3\text{CH}_2\text{CH}_3$ $\text{CH}_3\text{CH}_2\text{COCH}_2\text{CH}_3$
 $\text{CH}_3\text{CH}_2\text{OH}$

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3COCH₃ and CH₃CH₂CHO; [1] Accept full or condensed structural formulas. Ignore incorrect names as long as structures are correct. (b) same/similar (types of) bonds / both contain the carbonyl group/C=O; [1] Do not accept same functional group. (c) (mass spectrum of) CH₃CH₂CHO contains peak at $m/z = 29$ / CH₃COCH₃ does not contain peak at ...

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CH₃CH₂COOCH₃ or CH₃COOCH₂CH₃; this is because there are 3 peaks / 3:2:3 ratio; explanation of splitting into a singlet a triplet and a quartet; methyl propanoate/CH₃CH₂COOCH₃ is correct isomer because of higher chemical shift value of singlet (3.6 instead of 2.0-2.5); [4]

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